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Nucleosides, Nucleotides and Nucleic Acids

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Nucleosides. IL. Synthesis and Properties of 2,4-Quinazolinedione N-1-2' Deoxy-, 3'-Deoxy- and 2',3'-Dideoxynucleosides

Martin Dunkela; Wolfgang Pfleiderera

^a Fakultät für Chemie, Universität Konstanz, Konstanz, FRG

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NUCLEOSIDES. IL¹
SYNTHESIS AND PROPERTIES OF 2,4-QUINAZOLINEDIONE N-1-2'DEOXY-,
3'-DEOXY- AND 2',3'-DIDEOXYNUCLEOSIDES

Martin Dunkel and Wolfgang Pfleiderer
Fakultät für Chemie, Universität Konstanz
Postfach 5560, 7750 Konstanz, FRG

Abstract. A series of 6- and/or 7-substituted 2,4-quinazoline-dione N-1-deoxyribofuranosides have been synthesized and characterized. The 2'-deoxy-B-D-ribofuranosides 23-28 have been prepared by transformation of the corresponding ribofuranosides by chemical deoxygenation. Direct glycosidation to the β-anomers with a 2'-deoxyribofuranosyl donor to pure anomers failed due to missing diastereoselectivity and difficult separation of the reaction products. The synthesis of the 3'-deoxy-B-D-ribofuranosides 54-58, however, was achieved by glycosidation of the trimethylsilylated 2,4-quinazolinediones 43-47 with an appropriate 3'-deoxyribofuranosyl donor (48). The 2',3'-dideoxy-β-D-ribofuranosyl derivatives 63-66 were again obtained by chemical deoxygenation of the corresponding 2'-deoxy-B-D-nucleosides, since all experiments of direct glycosidation with a 2',3'-dideoxyribofuranosyl donor as well as the chemical conversion of the corresponding ribonucleosides into the 2',3'-dideoxynucleosides failed due to side reactions. The newly synthesized compounds have been identified by UV and ¹H-NMR spectra as well as elemental analyses.

In a recent paper we have described the chemical synthesis and physical properties of various quinazoline N-1 ribofuranosides, which can be regarded as benzo[d]uridine

Dedicated in memoriam to Tohru Ueda and in admiration of his scientific achievements.

and -cytidine analogues, respectively. So far, these nucleosides have not shown any biological activity, but there is the potential that the corresponding 2'- and 3'-deoxy, as well as the 2',3'-dideoxyribofuranosides, will reveal the anticipated activity. Since 2',3'-dideoxynucleosides belong to the most potent anti-HIV-1 agents efforts to prepare new types of structural analogues modified in the base moiety can be justified. We report about our studies to gain access to the various types of 2,4-quinazolinedione N-1- β -D-deoxyribonucleosides.

SYNTHESES

For the preparation of 2'-deoxy- β -D-ribofuranosides there are currently two general methods in use - the direct glycosylation procedure and the deoxygenation process of an appropriate precursor by radical substitution. The former approach reveals the disadvantage of low diastereoselectivity due to the lack of a 2'-O-acyl function, which controls, by neighbouring participation, glycoside bond formation according to the Tipson-Baker trans-rule. Some catalysts like ZnCl2 and CuI show, in the 2'-deoxyribonucleoside series, a positive effect towards a preferntial β -orientation between sugar and aglycone giving this method some chance of preparative application.

Our own experiments with various 2,4-quinazoline-diones following the basic work of R.K. Robins et al. ⁷ as well as by use of several catalysts in the glycosylation reaction of the silylated hetero-bases with 2-deoxy-3,5-di-0-p-toluoyl- α -D-erythro-pentofuranosyl chloride, unfortunately, gave no or little β -diastereoselectivity. The resulting anomeric mixtures of diastereomers were difficult to separate. Separation could only be achieved on small scale by MPLC or preparative TLC. From the physical data of pure 1-(2-deoxy- β -D-ribofuranosyl)-2,4-quinazolinedione (23) it can be concluded that the direct glycosylation of 2,4-quinazolinedione (43) by R.K. Robins et al. ⁷ did not lead to the pure β -ano-

mer, which has a melting point of $185-186^{\circ}$ and not $167-168^{\circ}$, as reported earlier. Also, the acylation of $\underline{23}$ by p-toluoyl chloride to $1-(2-\text{deoxy}-3,5-\text{di}-0-\text{p-toluoyl}-\beta-D-\text{ribofuranosyl})-2,4-quinazolinedione with a melting point of <math>147-148^{\circ}$ indicated that the previously described material (m.p. $129-134^{\circ}$) was not homogeneous.

Another general approach to 2'-deoxy- β -D-ribofuranosyl nucleosides has been demonstrated by the chemical transformations of the 2'-hydroxy group in the corresponding β -D-ribofuranosides utilizing the reductive removal of 2'-halo-and thiol substituents $^{8-23}$. Finally, Barton's radical deoxygenation process 24,25 of thio- and dithiocarbonates was successfully applied to nucleoside chemistry by M.J. Robins et al. 4 and became, in recent years, a standard procedure for the preparation of deoxy- and dideoxy nucleosides, in general.

When we applied the Robins method to 2,4-quinazolinedione ribofuranosides we encountered some difficulties upon the introduction of the phenoxythiocarbonyl function into the 3',5'-protected nucleosides 7-11, which could be prepared in nearly quantitative yields from the ribosides 1-5 by the Markiewicz reagent. Phenoxythiocarbonylation of 7 under normal conditions led to a mixture of the desired product 12 in 41 % yield, but also a second reaction product was identified as the 3',5'-blocked $0^2,2'$ -anhydronucleoside 17 and was isolated in 40 % yield. Another experiment showed that the anhydronucleoside 17 is formed quantitatively from 12 when treated with 4-dimethylaminopyridine in CH₃CN at room temp.. Variations of the conditions using phenoxythiocarbonyl chloride/pyridine in CH₂Cl₂ at 0°C gave a clean acylation of the 2'-OH group and 12 was isolated in 80 % yield. Reductive cleavage of the phenoxythiocarbonate 12 by tri-n-butylstannane in a radical chain reaction afforded 18 in 83 % yield, which led on subsequent deblocking of the Markiewicz protecting group with tetrabutylammonium fluoride (TBAF) to 1-(2-deoxy-β-D-ribofuranosyl)-2,4-quinazolinedione

 $(\underline{23})$ in 75 % isolated yield. The four step procedure $\underline{1} \rightarrow \underline{7} \rightarrow \underline{12} \rightarrow \underline{18} \rightarrow \underline{23}$ can also be performed sequentially without isolation and purification of the intermediates to give a good yield of the final product. In the same manner 6-methyl- $(\underline{24})$, 7-methyl- $(\underline{25})$, 6,7-dimethyl- $(\underline{26})$, and 6,7-dimethoxy-1- $(\underline{27})$, respectively, were synthesized in overall yields of 48-70 %.

In a second sequence of reactions the 2,4-quinazolinedione N-1-ribosides 1 and 2 were treated with Mattock's reagent²⁶ α-acetoxyisobutyryl bromide and acetyl bromide, respectively, as well as acetyl chloride to form the 2'-bromo-(33, 34) and 2'-chloro-3',5'-di-0-acetyl-ribonucleosides (35, 36). All attempts to remove the bromine by catalytic hydrogenation have failed, so far, and led to the formation of the corresponding 0^2 , 2'-anhydronucleosides which have, however, not been characterized further. The reduction of the haloatoms by tri-n-butylstannane worked well and gave good yields with the more stable 2'-chloro-2'-deoxyribonucleosides 35 and 36 affording in the radical chain reaction 37 and 38, respectively. In order to avoid the formation of peracylated byproducts in the transformations of 1 and 6 into the halo-derivatives acetyl chloride was replaced by the bulky pivaloyl chloride providing a much cleaner reaction to 39 and 40 in excellent yields. Tri-n-butylstannane reduction afforded crystalline 1-(2-deoxy-3,5-di-0-pivaloy1-β-D-ribofuranosyl)-2,4-quinazolinedione (41) and its 6-chloro derivative (42), both of which led by Zemplen's deacylation procedure 27 to the free 2'-deoxyribonucleosides 23 and 28, respectively. This second method can be regarded as an effective alternative to the Barton-Robins approach, which uses much more expensive reagents and is more time-consuming.

Our investigations have also been extended to the 3'-deoxynucleoside series, since this group of nucleosides $^{28-35}$ has attracted some attention after the isolation of cordycepin 36 in 1951. We decided to synthesize 2,4-quinazolinedione N-1-3'-deoxynucleosides $\underline{54-58}$ by direct glycosylation of the

heterobases $\underline{43}$ - $\underline{47}$ using Vorbrüggen's conditions 37 . The 2,4-quinazolinediones $\underline{43}$ - $\underline{47}$ were first silylated and then treated with 1,2-di-0-acetyl-3-deoxy-5-0-p-toluoyl- β -D-ribose 38 in CH₂Cl₂ under trimethylsilyl triflate catalysis to give in moderate to good yields the corresponding protected N-1-3'-deoxy- β -D-ribofuranosides $\underline{49}$ - $\underline{53}$. Deacylation according to Zemplen 27 by sodium methoxide led to the free 2,4-quinazolinedione N-1-3'-deoxy- β -D-ribofuranosides $\underline{54}$ - $\underline{58}$ in crystalline form.

Finally we have concentrated our efforts towards the synthesis of quinazoline-2',3'-dideoxynucleosides as poten-

A three step sequence of reactions starting from the 2,4-quinazolinedione-N-1-2'-deoxy- β -D-ribofuranosides 23-26 was finally successful in preparing the corresponding 2',3'-dideoxynucleosides 63-66. In the first place a selective monoacylation led to the 5'-O-p-toluoyl derivatives 29-32 in 65-75 % yield, which were then reacted at the 3'-OH group with phenoxythiocarbonyl chloride in pyridine and without isolation of the intermediates followed by subsequent reduction with tri-n-butylstannane/AIBN in toluene according to the Barton-Robins conditions 4,24,25 formed the $1-(2,3-dide-oxy-5-O-toluoyl-<math>\beta$ -D-ribofuranosyl)-2,4-quinazolinediones 59-62 in 72-99 % yield. Deacylation to the free quinazoline-2',3'-dideoxynucleosides 63-66 could be achieved in the usual manner by transesterification in methanol/sodium methoxide 27.

PHYSICAL PROPERTIES

Characterization of the newly synthesized quinazoline deoxynucleosides was based on elemental analyses, UV and $^1\mathrm{H-NMR}$ spectra. The site of attachment of the sugar moiety in the glycosylation reactions of the 2,4-quinazolinediones 43-47 to the corresponding 3'-deoxynucleosides 49-58 at the

Table 1 - Physical Data of Quinazoline N-1-Deoxyribonucleosides

	pKa	UV -	Abso	rption Sp	pectra		pH	Mole-
	in H ₂ 0	λ _{max} (nm)			log ε		Sol- vents	cular Form
7		217 [242]	306	4.63 [3	3.96]	3.60	Me0H	0
		219 [242]	315	4.62 [3	3.97]	3.57	MeOH	0
<u>8</u> <u>9</u>		222 253 [270]	309	4.66	4.10 [4.02]	3.67	МеОН	0
<u>10</u>		224 248 [268]	315	4.65	4.10 [3.95]	3.60	Me0H	0
<u>11</u>	ļ l	230 [257]	323	4.60	[3.80]	3.79	Me0H	0
12		237 [272] [281]	305	5.27 [4	4.07][3.92]	3.58	Me0H	0
<u>17</u>		221 [242] 297	308	4.49 [4	4.15] 3.71	3.64	Me0H	0
<u>18</u>		220 [240]	305	4.60 [3	3.90]	3.60	Me0H	0
<u>23</u>	9.92	218 [241] 221 [241] [260]	306 307	4.60 [3 4.64 [4	3.91] 4.08][3.58]	3.60 3.65	7.0 12.0	0 -
24	10.02	220 [244] 223 [262]	315 315	4.61 [3 4.65	3.98] [3.49]	3.60 3.63	7.0 12.0	0
<u>25</u>	9.89	224 [246] 225 [260]	306 306	4.62 [3 4.70	3.87] [3.74]	3.62	7.0 12.0	0 -
<u>26</u>	9.98	226 [248] [259] 226 [262]	313 312	4.63 [3 4.70	3.93] [3.76] [3.70]		7.0 12.0	0 -
<u>27</u>	10.31	235 [261] 230 [258] [269]	318 317	4.56 4.61 [3	[3.89] [3.66] [3.66]		8.0 13.0	0 -
28	9.35	221 244 [252] 224 [252]	317 317	4.64 4.63	4.05 [3.87] [3.90]		7.0 12.0	0 -
29		217 [236]	305	4.69 [4	4.40]	3.65	MeOH	0
<u>30</u>	Ì	220 [238]	313	4.66 [4	4.38]	3.58	Me0H	0
<u>31</u>		223 [238]	305	4.71 [4	4.35]	3.61	MeOH	0
32		224 [240]	311	4.78 [4	4.39]	3.68	MeOH	0
33		219 [240]	310	4.62 [3	3.98]	3.58	Me0H	0
<u>34</u>		218 [241]	309	4.61 [3	3.95]	3.60	MeOH	0
<u>35</u>		216 [240]	303	4.60 [3	3.87]	3.53	Me0H	0
<u>36</u>	[[218 [240]	310	4.62 [3	3.98]	3.58	MeOH	0
<u>37</u>		217 [240]	304	4.51 [3	3.81]	3.49	MeOH	0
<u>38</u>		219 [241]	314	4.63 [3	3.93]	3.60	MeOH	0

(continued)

Continuation: Table 1

	pKa	UV - Absor	ption Spectra	pH Sol-	Mole- cular
	in H ₂ 0	λ _{max} (nm)	log €	vents	Form
39		216 237 303	4.62 3.94 3.58	MeOH	0
40		219 244 [250] 314	4.61 4.02 [3.93] 3.48	MeOH	٥
41		217 [239] 304	4.64 [3.95] 3.62	Me0H	0
42		220 245 273 317	4.62 4.05 3.46 3.49	Me0H	0
49		217 [238] 306	4.63 [4.33] 3.54	MeOH	0
50		219 [237] [265] 302	4.71 [4.44] [3.68] 3.63	Me0H	0
51		223 [237] 302	4.68 [4.34] 3.60	MeOH	0
52		224 [238] 310	4.68 [4.38] 3.56	Me0H	0
53		235 [260] 317	4.68 [4.02] 3.80	MeOH	0
54	9.97	218 [239] 306 221 [240] [257] 307	4.59 [3.93] 3.59 4.63 [4.07][3.62] 3.63	7.0 12.0	o -
<u>55</u>	9.94	220 [241] 315 223 [260] 315	4.61 [4.01] 3.60 4.65 [3.57] 3.63	7.0 12.0	0
<u>56</u>	9.86	224 [246] [254] 306 225 [260] 306	4.64 [3.92] [3.85] 3.63 4.70 [3.74] 3.66	7.0 12.0	0 -
<u>57</u>	9.73	226 [248] [260] 314 226 [262] 312	4.63 [3.94] [3.75] 3.63 4.70 [3.70] 3.68	7.0 12.0	0 -
<u>58</u>	9.92	235 261 319 230 [258] [269] 317	4.57 3.90 3.86 4.63 [3.82][3.67] 3.89	7.0 12.0	0 -
<u>59</u>	}	217 [235] 306	4.61 [4.33] 3.57	MeOH	0
<u>60</u>	}	220 [237][271] 311	4.67 [4.41] [3.65] 3.65	Me0H	0
61	1	223 [238] [269] 304	4.74 [4.41] [4.05] 3.80	Me0H	0
62		225 [240] 311	4.68 [4.30] 3.58	MeOH	0
63	9.96	218 [240] 307 221 [240] [260] 307	4.59 [3.93] 3.59 4.63 [4.08] [3.54] 3.63	7.0 12.0	0 -
64	9.82	221 [244] 316 224 [260] 315	4.58 [3.96] 3.58 4.63 [3.51] 3.61	7.0 12.0	0 -
<u>65</u>	9.96	224 [246] [256] 306 225 [260] 306	4.61 [3.86] [3.79] 3.60 4.68 [3.69] 3.65	7.0 12.0	0 -
66	9.82	221 [244] 316 224 [260] 315	4.58 [3.96] 3.58 4.63 [3.51] 3.55	7.0 12.0	0 -

^[] Shoulder

N-1 ring atom is derived from UV spectral comparison with the structurally proven 1- β -D-ribofuranosyl-2,4-quinazoline-diones 1 . The deprotected deoxynucleosides 23-28, 54-58, and 63-66 were selected for pK_a determination and revealed expectedly very similar values (table 1). The monoanions of these compounds show only a small bathochromic shift of the long wavelength absorption band, which is typical for an N-1 rather than N-3 substitution.

The ¹H-NMR spectra of the various types of quinazoline-N-1-nucleosides reveal further information about conformational relations of the flexible parts of the molecules. The anomeric protons in the 3',5'-0-(1,1,3,3-tetraisopropyldisiloxane-1,3-diyl) (TIPDS) derivatives 7-11 appear as singlets analogous to 3',5'-TIPDS-uridine 60 and indicate a 3 E or Nconformation of the sugar-pucker. Furthermore the 2'- and 3'-protons resonate in these molecules at much lower field than in the uridine analogue pointing to a high population of syn-conformation of base and sugar moiety leading to a deshielding of the B-protons induced by the diamagnetic anisotropy of the 2-carbonyl function. The shift of the 3'-protons to lower field is then more pronounced in the N-conformer as stated. The same arguments also explain nicely the strong separation of the 2'- and 2"-proton signals in the 2,4-quinazolinedione N-1-(2-deoxy-B-D-ribofuranosides) 23-32. We assume that these derivatives prefer again the synconformation, wherein the 2'-deoxyribofuranoside moieties adapt the 2 E or S-pucker bringing the β -located 2'-H in close proximity to the 2-carbonyl group of the hetero-base.

The distinct chemical shift differences of the 3'- and 3"-proton signals in the quinazoline N-1-(3-deoxy- β -D-ribo-furanosides) $\underline{49}$ - $\underline{58}$ could be discussed under the same aspects pointing to an arrangement which indicates the presence of equal amounts of 2 E- and 3 E-conformers.

The 1 H-NMR spectra of the quinazoline N-1(2,3-dideoxy- β -D-nucleosides) $\underline{59}$ - $\underline{66}$ allow, so far, no concrete conformational assignments due to the complex patterns of the proton signals, which have not yet been analyzed in detail.

EXPERIMENTAL

¹H-NMR spectra were measured with a Bruker WM 250 and a Bruker AC 250 spectrometer with tetramethylsilane as an internal standard and on a δ-scale in ppm. UV-spectra were recorded on a Perkin-Elmer spectrophotometer Lambda 5. The pK-values were determined spectrophotometrically ⁶¹. Thin layer chromatography was performed on silica-gel sheets F 1550 LS 254 of Schleicher & Schüll. Silica Gel 60 (Merck) was used for ordinary column chromatography. Flash chromatography was done on Flash silica gel (Baker) according to their instructions. Drying of substances was achieved at 100°C or in a Büchi T0-50 drying oven under vacuum at 40°C. Melting points were measured with a Gallenkamp melting point apparatus and are not corrected.

 $\frac{1-[3,5-0-(1,1,3,3-\text{tetraisopropyldisiloxane-1,3-\text{diyl})-}{\text{B-D-ribofuranosyl}]-2,4-\text{quinazolinediones}} \; (7-11). \; \text{General}}{\text{procedure.}} \; - \; \text{A mixture of the 1-$\text{B-D-ribofuranosyl}-2,4-} \\ \text{quinazolinedione} \; (20 \text{ mmole}) \; (\underline{1-5}) \; \text{and} \; 1,3-\text{dichloro-1,1,3,3-} \\ \text{tetraisopropyldisiloxane} \; (5.9 \text{ ml,} \; 22 \text{ mmole}) \; \text{in pyridine}} \\ (100 \text{ ml}) \; \text{was stirred overnight at room temperature. After} \\ \text{evaporation the residue was partitioned between H}_20 \; (250 \text{ ml}) \\ \text{and ether} \; (500 \text{ ml}). \; \text{The organic layer was washed subsequent-} \\ \text{ly with 1N HCl} \; (250 \text{ ml}), \; \text{saturated NaHCO}_3 \; \text{solution} \; (250 \text{ ml}) \\ \text{and saturated NaCl solution} \; (250 \text{ ml}). \; \text{After drying over Na}_2-50_4 \; \text{the solution was filtered and evaporated in vacuum to} \\ \text{give an amorphous solid foam. The products are homogeneous} \\ \text{and do not afford chromatographical purification.}$

 $\frac{1-[3,5-0-(1,1,3,3-tetraisopropyldisiloxane-1,3-diyl)-}{\beta-D-ribofuranosyl]-2,4-quinazolinedione} \begin{picture}(7). - Yield 10.5 g (98 \%). Amorphous yellowish solid. Column chromatography with CHCl<math>_3$ /MeOH 49:1 gave 85 % colourless material. R = 0.79 (CHCl $_3$ /MeOH 19:1).

Anal. calc. for $C_{25}H_{40}N_2O_7Si_2$ (536.8): C, 55.94; H, 7.51; N, 5.22. Found: C, 55.58; H, 7.51; N, 5.15.

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Table 2 - ¹H-NMR Data of Quinazoline N-1-Deoxyribofuranosides

		- -	A Spec	tra in	05M0-90	H-NMR Spectra in D ₆ -DMSO or CDCl ₃ *	(6-valu	(6-values in ppm against TMS)	ainst TMS)						
	Ŧ	1'-H J1;2'	1,2	2'-H	1."2	3,-∓	3"-∓	4'-H 5'-H	H 5"-H	Aromatic H	tic H			Subst.	
*~!	8.14	5.94		4.79d		5.03t		3.97m 4.06m(2H)	5m(2H)	7.29t	7.29t 7.47d 7.68t	7.68t	8.18dd		
* ∞	8.26	5.95		4.78d		5.02t		3.95m 3.95	3.95m 3.95-4.09m(2H) 7.35d	7.35d	7.47dd 7.97d	7.97d		2.40s(3H)	
* o ;	8.45	5.91		4.81d		5.05t		4.04m(3H)	₽	7.10d	7.10d 7.25s	8.06d		2.47s(3H)	
* 2	8.17	5.90s		4.81d		5.05t		4.02m(3H)	Ŧ	7.23d	7.23d 7.90s			2.38s(3H)	2.30s(3H)
*={	8.41	5.82s		4.83d		5.05t		4.05m(3H)	~	6.84s	6.84s 7.53s			3.94s(3H)	3.98s(3H)
"≃	8.53	6.40s		6.62d		5.24dd		4.05m(3H)	⊋	7.24m(2H)	(H)	7.69t	8.19d		
<u>'≃</u> i		6.344	6.4	4.60m		5.43dd		4.05m(3H)	₽	7.30m(2H)	(ZH)	7.63t	8,12d		
<u>.</u> ∞	8.97	7.69dd		2.86m	2.31m	4.96m		3.86m 4.08m(2H)	N(2H)	7.21t	7.21t 7.55d	7.65t	8.18d		
ଅ	11.23	6.76t		2.60m	1.90m	4.33m		3.69m(3H)	₽	7.29t	7.29t 7.64t	7.85d	8.014		
24	11.57	6.65t		2.61m	1.92m	4.37m		3.65m(3H)	÷	7.77m(2H)	(ZH)	7.46d		2.33s(3H)	
গ্ৰ	11.54	6.69t		2.62m	1.91m	4.42m		3.68m(3H)	₽	7.10d	7.10d 7.77s	7.87d		2.37s(3H)	
81	11.47	6.69t		2.60m	1.90m	4.43m		4.11m 3.68m(2H)	(SH)	7.72s	7.75s			2.23s(3H)	2.29s(3H)
2	11.51	6.71t		2.65m	1.91m	4.42m		3.66m(3H)	÷	7.16s	7.16s 7.38s			3.80s(3H)	3.88s(3H)
8	11.80	6.16t		2.57m	1.91m	4.37m		3.66m(3H)	÷	7.66dd	7.66dd 7.92m(2H)	ZH)			
ଛା	11.66	6.68t	7.1	2.76m	2.05m	3.99™		4.61-4.45m(3H)	(HE)	7.23m	7.23m 7.37m 7.59d	7.59d	7.98d		
ଛା	11.59	6.66t		2.72m	2.03⋒	3.92m		4.58m 4.66dd 4.43dd	dd 4.43dd	7.06dd	7.06dd 7.45d	7.77d		2.25s(3H)	
티	11.54	6.60t		2.78⋒	2.08≡	4.01m		4.54m(2H)	4.43dd	7.09d	7.42s	7.86		2.31s(3H)	
	11.48	6.59t		2.76m	2.10m	4.01m		4.55m(2H)	4.43m	7.39s	7.76s			2.23s(6H)	
[33]	8,96	6.55d	4.9	5.45m(2H)	(£			4.36m(2H)	4.53dd	7.34t	•	7.49d 7.69t	8.25dd		
3g	8.95	6.47d	5.1	5.36m(2H)	£			4.26m(2H)	4.32dd	7.31t	7.41dd 7.97d	7.97d		2.35s(3H)	
	11.82	6.45d	5.0	5.42m(2H)	Œ			4.31m(3H)	=	7.35t	7.70m(2H)	ZH.)	8.04d		
<u>اچ</u>	8.8 %	6.30d	5.0	5.31dd	5.4	5.48t		4.29m 4.23dd 4.45dd	dd 4.45dd	7.29d	7.41dd		7.97dd	2.35s(3H)	
37	11.67	6.64t	7.3	2.93⋒	2.20m	5.35m		4.15m 4.31	4.31m(2H)	7.30t	7.70m(2H)	(표	8.02d		
'ස <u> </u>	9.54	6.72t	7.6	3.06m	2.18m	5.40m		4.18m 4.38	4.38d 4.46dd	7.39dd	7.39dd 7.51d		8.00d	2.35s(3H)	
$\frac{1}{2}$															

Table 2 - Continuation

				-												
				H-NMR S	pectra	10 06-04	$ ext{H-NMR}$ Spectra in $ extstyle{06} extstyle{06} extstyle{06} extstyle{06}$ (6-values in ppm against TMS)	3 (6-va	lues in	ppm aga	inst T	€ Se				
	₹		J1,21	2Н	2"-H	3,-₩	3"-н	±.4	₽-,5	5"-#	Aromatic H	1c H			Subst.	
39*	8.96	6.37d	4.8	5.40dd		5.53t		4.3	4.37m(3H)		7.34t	7.45d	.34t 7.45d 7.69t	8.24dd		
육	9.08	6.30d		5.40dd		5.50m		4.3	4.35m(3H)		7.40d	7.40d 7.65dd		8.20dd		
‡∓	8.60	6.71t	7.6	3.11m	2.18m	5.40m		4.18m 4.35dd 4.48dd	4.35dd	4.48dd	7.31t	7.31t 7.62m(2H)	2H)	8.22dd		
4 2	9.50	6.64t	7.5	3.02m	2.16m	5.36m		4.25m '	.25m 4.30dd 4.47dd	4.47dd	7.54m(2H)	2H)		8.17d		
49	9.11	6.184	5.9	5.82dd		2.88m	2.19dd	4.65m(2H)		4.48dd	7.25m(2H)	2H)		8.21d		
ଜା	8.97	6.144	3.0	5.78dd		2.84m	2.15dd	4.64m(2H)	•	4.44dd	7.25m(2H)	2H)		7.96d		
21*	8.82	6.00d	2.3	5.89dd		2.87m	2.19dd	4.61m(2H)		4.46dd	7.09d	7.09d 7.18s		8.08d		
22	8.46	6.01d	2.5	5.88dd		2.84m	2.23dd	4.61m(2H)		4.46m		7.15s		7.93	2.33s(3H)	2.28s(3H)
23 [*]	9.10	5.99d	2.5	5.90dd		2.85⋒	2.20dd	4.64m(2H)		4.45m	6.87s	6.87s 7.57s			3.95s(3H)	3.90s(3H)
ক্র	11.67	6.12d	5.0	4.62m		2.35m	1.83m	4.20m	.20m 3.49m(2H)	£	7.29m	7.29m 7.68m(2H)	2H)	8.01d		
िक्ष	11.61	P60.9	4.9	4.61m		2.34m	1.82m	4.19m	3.54m(2H)	£	7.48dd	.48dd 7.59d 2.81d	2.81d		2.34s(3H)	
8	11.57	6.13d	5.0	4.64m		2.43m	1.84m	4.22m	3.61m 3.50m	3.50m	7.10d	7.62s	7.88d		2.43s(3H)	
22	11.50	6.12d	5.1	4.62m		2.40m	1.83m	4.21m 3	3.63dd 3.50dd	3.50dd	7.60s	7.74s			2.34s(3H)	2.24s(3H)
8	11.54	6.20d	5.1	4.64m		2.41≡	1.79m	4.15m 3	3.67m 3.50m	3.50m	7.02s	7.38s			3.80s(3H)	3.88s(3H)
65	11.65	6.57t		2.	2.10 -	2.50 m (4H)	4H)	4.39m 4	4.59dd 4.43dd	4.43dd	7.23t	7.43t	7.61d	P66.7		
ଞା	11.57	6.55t		2.	2.10 -	2.50 m (4H)	4H)	4.30m 4	4.63dd 4.40dd	4.40dd	7.10dd	7.10dd 7.47d	7.78s		2.26s(3H)	
61	8.78	6.53t		2.:	2.20 -	2.65 m (4H)	(₩	4.41m 4	4.64dd 4.50dd	4.50dd	7.08d	7.38s		8.08d	2.38s(3H)	
29	11.47	6.50t		2.10	5	2.50 m (4H)	(₩	4.32m 4	4.52dd 4.42dd	4.42dd	7.38s	7.74s			2.23s(6H)	
8	11.50	6.55t		-	- 06.1	2.35 m (4H)	4H)	3.99m 3	3.61m(2H)	3	7.23m	7.61m	7.72m	7.98d		
8	11.55	6.50t		-	- 06.1	2.35 m (4H)	4H)	3.96m 3	3.60m(2H)	£	7.49dd	.49dd 7.68d	7.80s		2.34s(3H)	
92	11.52	6.54t		-	- 06.1	2.35 m (4H)	4H)	3.98m 3	3.69m	3.55m	7.10d	7.70s	7.88d		2.39s(3H)	
8	11.48	6.52t		-	- 06.1	2.40 m (4H)	4H)	3.97m 3	3.66m 3.57m	3.57⋒	7.69s	7.74s			2.24s(6H)	

s = Singlet; d = doublet; dd = doublet of doublet; t = triplet; m = multiplet.

- $\frac{6-\text{Methyl-1-[3,5-0-(1,1,3,3-tetraisopropyldisiloxane-1,3-diyl)-\beta-D-ribofuranosyl]-2,4-quinazolinedione}{1,3-diyl)-\beta-D-ribofuranosyl]-2,4-quinazolinedione} (8). Yield 11.0 g (99 %). Amorphous yellowish foam. R_f = 0.86 (CHCl₃/MeOH 19:1).$
- Anal. calc. for $C_{26}H_{42}N_2O_7Si_2$ (550.8): C, 56.70; H, 7.69; N, 5.09. Found: C, 56.40; H, 7.70; N, 5.07.
- $\frac{7-\text{Methyl-1-[3,5-0-(1,1,3,3-tetraisopropyldisiloxane-1,3-diyl)-\beta-D-ribofuranosyll-2,4-quinazolinedione}{1,3-diyl)-\beta-D-ribofuranosyll-2,4-quinazolinedione} (9). -Yield 11.0 g (99 %). Amorphous solid. <math>R_f = 0.88$ (CHCl₃/MeOH 19:1).
- $\frac{\text{Anal}}{\text{Anal}}$. calc. for $\text{C}_{26}^{\text{H}}_{42}^{\text{N}}_{2}^{\text{O}}_{7}^{\text{Si}}_{2}$ (550.8): C, 56.70; H, 7.69; N, 5.09. Found: C, 56.36; H, 7.69; N, 5.09.
- $\frac{6,7-\text{Dimethyl-1-[3,5-0-(1,1,3,3-tetraisopropyldisiloxane-1,3-diyl)-\beta-D-ribofuranosyl]-2,4-quinazolinedione}{1,3-diyl)-\beta-D-ribofuranosyl]-2,4-quinazolinedione} (10). Yield 11.06 g (98 %). Amorphous solid. <math>R_f = 0.75$ (CHCl₃/MeOH 19:1).
- Anal. calc. for $C_{27}H_{44}N_2O_7Si_2$ (564.8): C, 57.42; H, 7.85; N, 4.96. Found: C, 56.92; H, 8.16; N, 4.87.
- $\frac{6.7-\text{Dimethoxy-1-[3,5-0-(1,1,3,3-tetraisopropyldisilo-xane-1,3-diyl)-\beta-D-ribofuranosyl]-2,4-quinazolinedione}{\text{Yield 11.32 g (95 %). Amorphous yellowish foam. R}_{f}=0.67}$ (CHCl $_{3}$ /MeOH 19:1).
- Anal. calc. for $C_{27}H_{44}N_2O_9Si_2$ (596.8): C, 54.33; H, 7.43; N, 4.69. Found: C, 54.12; H, 7.38; N, 4.50.
- $\frac{1-[2-0-Phenoxythiocarbonyl-3,5-0-(1,1,3,3-tetraisopro-pyldisiloxane-1,3-diyl)-\beta-D-ribofuranosyll-2,4-quinazoline-dione (12). (a) To a solution of 7 (1.07 g, 2 mmole) in CH₃CN (30 ml) were added 4-dimethylaminopyridine (DMAP) (0.38 g, 4.1 mmole) and phenoxythiocarbonyl chloride (0.3 ml, 2.2 mmole) with stirring. After reaction for 16 h at room temperature the solution was evaporated and the residue par-$

titioned between $\rm H_2O$ (120 ml) and $\rm CHCl_3$ (150 ml). The organic layer was dried over $\rm Na_2SO_4$, filtered and concentrated to a small volume for chromatography on a silica-gel column (4.5x9 cm) with $\rm CHCl_3$. The main fraction gave on evaporation a yellowish oil, which could be recrystallized from $\rm CH_3CN$ to give colourless crystals. Yield 0.54 g (40 %), m.p. 112-113°C. $\rm R_f=0.71$ (CHCl $_3/\rm MeOH$ 99:1).

(b) The same reaction was performed in $\mathrm{CH_2Cl_2}$ (10 ml) at 0°C for 5 h, followed by stirring at room temperature overnight. Work-up and purification according to (a) gave an amorphous solid. Yield 1.15 g (85 %).

 $\frac{\text{Anal}}{\text{calc.}}$ for $C_{32}H_{44}N_2O_8SSi_2$ (672.9): C, 57.11; H, 6.59; N, 4.16. Found: C, 57.08; H, 6.66; N, 4.50.

 $\frac{0^2,2'-\text{Anhydro-1-[3,5-0-(1,1,3,3-tetraisopropyldisilo-xane-1,3-diyl)-\beta-D-arabinofuranosyl]-2,4-quinazolinedione}{(\underline{17}).}$ (a) Continued elution of the column in $\underline{7}$ a) led to a second fraction, which gave on evaporation a solid foam. Yield 0.43 g (41 %). $R_f = 0.32$ (CHCl₃/MeOH 99:1).

 $\frac{\text{Anal}}{\text{Anal}}$. calc. for $\text{C}_{25}\text{H}_{38}\text{N}_2\text{O}_6\text{Si}_2$ (518.8): C, 57.88; H, 7.38; N, 5.40. Found: C, 57.90; H, 7.40; N, 5.28.

- (b) A solution of $\underline{12}$ (0.67 g, 1 mmole) in CH $_3$ CN (15 ml) was treated with 4-dimethylaminopyridine (0.244 g, 2 mmole) at room temperature overnight. It was evaporated, the residue extracted with CHCl $_3$ (30 ml) and H $_2$ O (20 ml). The organic phase was dried over Na $_2$ SO $_4$ and gave on further evaporation an amorphous solid chromatographically identical with fraction a). Yield 0.5 g (95 %).
- $\frac{1-[2-Deoxy-3,5-0-(1,1,3,3-tetraisopropy]disiloxane-1,3}{diyl)-\beta-D-erythro-pentofuranosyl]-2,4-quinazolinedione} (18).$ A solution of 12 (1.91 g, 2.84 mmole) in toluene (30 ml) was treated with argon for 20 min, then tri-n-butylstannane (2.26 ml, 8.52 mmole) and azo-bis-(isobutyronitrile) (AIBN) added and the mixture heated to 70°C under argon atmosphere for 3 h. After cooling ether (200 ml) and 0.1 N NaOH (100 ml)

were added and shaken. The organic phase was dried over Na_2 - SO_4 , filtered and evaporated. The oily residue was purified by column chromatography in CHCl $_3$. The main fraction gave on evaporation an amorphous solid. Yield 1.23 g (83 %).

Anal. calc. for $C_{25}H_{40}N_2O_6Si_2$ (520.8): C, 57.65; H, 7.74; N, 5.38. Found: C, 57.40; H, 7.60; N, 5.31.

<u>1-(2-Deoxy-β-D-erythro-pentofuranosyl)-2,4-quinazoline-</u> dione (23). - (a) A mixture of 2,4-quinazolinedione (43) (2.93 g, 18.1 mmole), a few crystals of ammonium sulfate and acetamide were refluxed in hexamethyldisilazane (HMDS) (70 ml) for 18 h under exclusion of moisture. Excess of HMDS was removed in vacuo, the residue dissolved in CHCl₂ (400 ml), then 2-deoxy-3,5-di-0-p-toluoyl-D-erythro-pentofuranosyl chloride (6.61 g, 18.1 mmole) and CuI (3.3 g) added and stirred at room temperature for 20 h. The reaction mixture was treated with saturated $NaHCO_3$ solution, the organic layer separated and the aqueous phase again extracted twice with CHCl₃. The combined extracts were dried over Na_2SO_A , evaporated and purified by column chromatography on silicagel first with $CHCl_3$ and then with $CHCl_3/ethyl$ acetate 4:1. The main fraction was evaporated and the residue recrystallized from EtOH to give a 3:1 α , β -mixture of 1-(3,5-di-0-ptoluoyl-2-deoxy-α,β-D-erythro-pentofuranosyl)-2,4-quinazolinedione. Yield 7.71 g (80 %). Separation of the anomers could be achieved by chromatography on preparative plates (40x20x0.2 cm) in toluene/EtOAc 4:1. The lower moving band gave pure 1-(3,5-di-0-p-toluoy1-2-deoxy-B-D-erythro-pentofuranosyl)-2,4-quinazolinedione. Its structure was proven by acylation of 23 (0.56 g, 2 mmole) with p-toluoyl chloride (0.53 ml, 5 mmole) in pyridine (10 ml) by stirring at 0°C for 2 h and 20 h at room temperature. The reaction mixture was poured on ice, extracted twice with CHCl_3 (50 ml), the organic phase dried over Na_2SO_4 and evaporated to dryness. The residue gave on recrystallization from benzene/cyclohexane colourless crystals. Yield 0.73 g (71 %), m.p. 147-148°C, lit. 7 129-134°C.

 $\frac{\text{Anal}}{\text{Anal}}$. calc. for $\text{C}_{29}\text{H}_{26}\text{N}_{20}$, (514.5): C, 67.70; H, 5.09; N, 5.44. Found: C, 67.37; H, 5.10; N, 5.70.

- (b) A solution of $\underline{18}$ (1.23 g, 2.36 mmole) and tetrabutylammonium fluoride (1.64 g, 5.2 mmole) in tetrahydrofuran (THF) (20 ml) was stirred for 1 h at room temperature. The precipitate was filtered, washed twice with ether. The solid was dissolved in boiling H_2O (8 ml), neutralized by AcOH and then more H_2O added till a clear solution on heating was obtained. On cooling colourless needles separated. Yield 0.48 g (75 %), m.p. $185-186^{\circ}C$, lit. 7 $167-168^{\circ}C$.
- (c) In a 0.2 N solution of sodium methoxide in MeOH (150 ml) 37 (12.5 g, 28 mmole) was stirred overnight at room temperature. After evaporation the residue was partitioned between H₂0 (50 ml) and ether (200 ml). The organic phase was washed with H₂0 (20 ml) and again separated. The combined aqueous layers were neutralized by AcOH and chilled to give colourless needles. Yield 6.66 g (85 %), m.p. 185°C.
- (d) In CH_2Cl_2 (200 ml) was dissolved $\frac{7}{2}$ (10.7 g, 20 mmole) and pyridine (2.77 g, 35 mmole). After cooling to 0°C phenoxythiocarbonyl chloride (6.04 g, 35 mmole) was added and the mixture stirred at 0°C for 5 h and then at room temperature for 18 h. After evaporation the residue was taken up in H₂0 (350 ml) and ether (500 ml). The eorganic layer was washed with 1 N HCl (200 ml), saturated $NaHCO_3$ solution (200 ml) and saturated NaCl solution (200 ml) and then dried over Na_2SO_4 . On evaporation, crude 12 was obtained. It was dissolved in toluene (300 ml), the solution purged with argon, tri-n-butylstannane (17.5 g, 60 mmole) and AIBN (0.3 g) added and then heated to 75°C for 3 h with stirring under argon atmosphere. After cooling it was diluted with ether (400 ml), washed with 0.1 N NaOH (250 ml) and phosphate buffer pH 7 (300 ml). The organic layer was dried over Na_2SO_A , filtered, evaporated to dryness, the residue (18) dissolved in THF (200 ml) and after addition of tetrabutylammonium fluoride (14.2 g, 45 mmole) the solution stirred for 1 h.

The precipitate was collected, washed with ether, dissolved in hot $\rm H_2O$ and then neutralized with AcOH. After cooling, the solid was filtered and then recrystallized from $\rm H_2O$ to give colourless crystals. Yield 6.66 g (85 %), m.p. 185°C.

Anal. calc. for $C_{13}H_{14}N_2O_5$ (278.3): C, 56.11; H, 5.07; N, 10.07. Found: C, 56.03; H, 5.17; N, 10.07.

6-Methyl-1-(2-deoxy-β-D-erythro-pentofuranosyl)-2,4-quinazolinedione (24). - Analogous to the preceding procedure from 8 (11.1 g; 20 mmole) via the intermediates 13 and 19. Yield 4.11 g (70 %), m.p. 230°C (decomp.).

Anal. calc. for $C_{14}H_{16}N_{2}O_{5}$ (292.3): C, 57.53; H, 5.52; N, 9.58. Found: C, 57.07; H, 5.53; N, 9.45.

 $\frac{7-\text{Methyl-1-}(2-\text{deoxy-}B-D-\text{erythro-pentofuranosyl})-2,4-}{\text{quinazolinedione}}$ (25). - According to procedure $\frac{7}{2}$ d) from $\frac{9}{3.54}$ g (60 %), m.p. 220°C (decomp.).

Anal. calc. for $C_{14}H_{16}N_{2}O_{5}$ (292.3): C, 57.53; H, 5.52; N, 9.58. Found: C, 57.21; H, 5.56; N, 9.53.

6,7-Dimethyl-1-(2-deoxy- β -D-erythro-pentofuranosyl)-2,4-quinazolinedione (26). - According to procedure 7 d) from 10 (11.3 g, 20 mmole) via the intermediates 15 and 21. Yield 3.64 g (62 %), m.p. 230°C (decomp.).

 $\frac{\text{Anal}}{\text{N}}$. calc. for $\text{C}_{15}\text{H}_{18}\text{N}_2\text{O}_5$ (306.3): C, 58.82; H, 5.92; N, 9.15. Found: C, 58.51; H, 5.98; N, 9.14.

 $\frac{6.7-\text{Dimethoxy-1-}(2-\text{deoxy-}\beta-\text{D-erythro-pentofuranosyl})-2.4-\text{quinazolinedione}}{2.4-\text{quinazolinedione}} \, (27). - \text{Analogous to procedure } \frac{7}{2} \, d)$ from $\frac{11}{2} \, (11.9 \, \text{g}, 20 \, \text{mmole})$ via the intermediates $\frac{16}{2} \, \text{and} \, \frac{22}{2}.$ Yield $\frac{3.17}{2} \, \text{g} \, (49 \, \%)$, m.p. $\frac{220}{2} \, \text{C} \, (\text{decomp.}).$

Anal. calc. for $C_{15}H_{18}N_2O_7$. H_2O (356.3): C, 50.56; H, 5.66; N, 7.86. Found: C, 51.01; H, 5.57; N, 7.88.

6-Chloro-1-(2-deoxy-β-D-erythro-pentofuranosyl)-2,4- quinazolinedione (28). - (a) In a 0.01 M solution of sodium methoxide in MeOH (20 ml) was dissolved 40 (0.75 g, 1.56 mmole) and the solution was stirred at room temperature overnight. After evaporation, the residue was treated with H_2 0 (10 ml) and ether (50 ml). The organic phase was washed again with H_2 0 (5 ml) and then the combined aqueous phases neutralized by AcOH. On cooling, colourless crystals were obtained. Yield 0.313 g (64 %), m.p. 220°C (decomp.).

 $\underline{\text{Anal}}$. calc. for $C_{13}H_{13}C1$ N_2O_5 (312.7): C, 49.93; H, 4.19; N, 8.96. Found: C, 49.95; H, 4.34; N, 8.99.

(b) A mixture of 6-chloro-2,4-quinazolinedione (4.33 g, 22 mmole), a few crystals of ammonium sulfate and acetamide was refluxed in 80 ml HMDS under anhydrous conditions for 15 h. It was evaporated under vacuum, the residue dissolved in $\mathrm{CH_2Cl_2}$ (100 ml), and then 2-deoxy-3,5-di-0-p-toluoyl- α -D-erythro-pentofuranosyl chloride (8.54 g, 22 mmole) added. The solution was stirred at room temperature for 24 h, evaporated and the residue recrystallized from EtOH to give a 1:1 mixture of 6-chloro-1-(2-deoxy-3,5-di-0-p-toluoyl- α , β -D-erythro-pentofuranosyl)-2,4-quinazolinedione. Yield 8.13 g (67 %). A part of this mixture (0.75 g) was separated into the pure components by medium pressure chromatography 62 on silica-gel in toluene/EtOAc 19:1. The faster migrating fraction gave on evaporation a crystalline solid of the β -anomer. Yield 0.24 g (32 %), m.p. 199°C.

Anal. calc. for $C_{29}H_{25}C1\ N_2O_7\ (549.0)$: C, 63.45; H, 4.59; N, 5.10. Found: C, 63.28; H, 4.58; N, 5.13.

The second fraction consisted of the α -anomer. Yield 0.28 g (37 %), m.p. 187°C.

 $\frac{\text{Anal}}{\text{Anal}}$. calc. for $C_{29}H_{25}C1$ N_2O_7 (549.0): C, 63.45; H, 4.59; N, 5.10. Found: C, 63.20; H, 4.65; N, 5.13.

The 6-chloro-1-(2-deoxy-3,5-di-0-p-toluoyl- β -D-erythropentofuranosyl)-2,4-quinazolinedione (0.55 g, 1 mmole) was treated analogously to procedure (a) to give colourless crystals. Yield. 0.225 g (72 %), m.p. 220°C (decomp.).

 $\frac{1-(2-\text{Deoxy}-5-0-\text{p-toluoyl}-\beta-\text{D-erythro-pentofuranosyl})-2,4-\text{quinazolinediones}}{2,4-\text{quinazolinediones}} (29-32). \text{ General procedure.} - \text{In pyridine}} (30 \text{ ml}) \text{ was dissolved the } 1-(2-\text{deoxy}-\beta-\text{D-ribofuranosyl})-2,4-\text{quinazolinedione}} (23-26) (10 \text{ mmole}) \text{ and after cooling to } 0^{\circ}\text{C} \text{ p-toluoyl chloride}} (1.45 \text{ ml}, 11 \text{ mmole}) \text{ added.}$ The mixture was stirred at 0°C for 3 h and at room temperature for 15 h. After dilution with ice-water (200 ml) the solution was extracted twice with EtOAc (80 ml). The organic extract was dried over Na_2SO_4 , evaporated, the oily residue dissolved in CH_2Cl_2 (75 ml) and then the boiling solution further diluted by slow addition of n-hexane (75 ml) forming a colourless precipitate. The crystalline material was collected after cooling.

 $\frac{1-(2-Deoxy-5-0-p-toluoyl-\beta-D-erythro-pentofuranosyl)-}{2,4-quinazolinedione} (29). - Yield 2.81 g (71 %), m.p. 180°C. R_f = 0.42 (CHCl₃/MeOH 19:1).$

Anal. calc. for $C_{21}H_{20}N_{2}O_{6}$ (396.4): C, 63.63; H, 5.09; N, 7.07. Found: C, 63.65; H, 5.20; N, 7.37.

 $\frac{6-\text{Methyl-1-}(2-\text{deoxy-5-0-p-toluoyl-}\beta-\text{D-erytho-pentofura-nosyl})-2,4-\text{quinazolinedione}}{\text{m.p.}\ 184-187°C.} R_f = 0.79 (CH_2Cl_2/\text{MeOH 9:1}).$

Anal.calc. for $C_{22}H_{22}N_2O_6$. 0.5 H_2O (419.4): C, 63.00; H, 5.53; N, 6.68. Found: C, 62.95; H, 5.42; N, 6.39.

 $\frac{7-\text{Methyl-1-}(2-\text{deoxy-5-0-p-toluoyl-}\beta-D-\text{erythro-pentofu-ranosyl})-2,4-\text{quinazolinedione}}{\text{m.p. }193-195^{\circ}\text{C. }R_{f}=0.58\text{ (CHCl}_{3}/\text{MeOH }19:1).}$

Anal. calc. for $C_{22}H_{22}N_2O_6$ (410.4): C, 64.78; H, 5.40; N, 6.82. Found: C, 64.84; H, 5.68; N, 6.56.

- $\frac{6.7-\text{Dimethyl-1-}(2-\text{deoxy-5-0-p-toluoyl-}\beta-\text{D-erythro-pento-furanosyl})-2.4-\text{quinazolinedione}}{\text{m.p. }199^{\circ}\text{C. }R_{f}=0.40\text{ (CHCl}_{3}/\text{MeOH 9:1}).}$
- Anal. calc. for $C_{23}H_{24}N_2O_6$ (442.5): C, 62.43; H, 5.92; N, 6.33. Found: C, 62.81; H, 5.89; N, 7.25.
- $\frac{1-(3,5-\text{Di-}0-\text{acetyl-}2-\text{bromo-}2-\text{deoxy-}\beta-\text{D-erythro-pentofu-ranosyl})-2,4-\text{quinazolinedione}}{(3.94\text{ g, 10 mmole})\text{ in CH}_3\text{CN}~(150\text{ ml})\text{ was heated to }50^\circ\text{C}\text{ and then a solution of acetyl bromide}~(4.1\text{ ml})\text{ in CH}_3\text{CN}~(15\text{ ml})}$ added dropwise within 15 min with stirring. The reaction mixture was refluxed for 2.5 h, then evaporated and the residue treated with H $_2$ 0 (100 ml) and CHCl $_3$ (2x100 ml). The organic phase was dried over Na $_2$ SO $_4$, filtered, evaporated and the resulting residue purified by flash chromatography (80 g silica-gel) in CHCl $_3$ /MeOH 49:1. The main fraction gave on evaporation an amorphous solid. Yield 2.6 g (59 %).
- (b) A mixture of $\underline{1}$ (0.59 g, 2 mmole), LiBr (0.52 g) and acetyl bromide (0.44 ml) in CH $_3$ CN (20 ml) was boiled under reflux for 1 h. Work-up and purification was performed analogously to (a). Yield 0.74 g (84 %).
- (c) Compound 1 (0.59 g, 2 mmole) was heated under reflux with α -acetoxyisobutyryl bromide (0.88 ml) in CH $_3$ CN (20 ml) for 2 h. Work-up and purification according to (a) gave an amorphous foam. Yield 0.45 g (51 %). R $_f$ = 0.82 (CHCl $_3$ /MeOH 19:1).
- $\frac{\text{Anal}}{\text{Anal}}$. calc. for $\text{C}_{17}\text{H}_{17}\text{BrN}_2\text{O}_7$ (441.2): C, 46.28; H, 3.88; N, 6.35. Found: C, 46.15; H, 3.95; N, 6.09.
- $\frac{6-\text{Methyl-1-(3,5-di-0-acetyl-2-bromo-2-deoxy-}\beta-D-erythro-pentofuranosyl)-2,4-quinazolinedione (34).}{2 (1.54 g, 5 mmole) in CH₃CN (70 ml) was heated to 50°C and then a mixture of acetyl bromide (3.7 ml, 30 mmole) in$

CH $_3$ CN (10 ml) added dropwise within 15 min. After refluxing for 2 h, the solution was evaporated, the residue treated with H $_2$ O (70 ml) and CHCl $_3$ (2x80 ml). The organic phase was dried over Na $_2$ SO $_4$ and then again concentrated to a syrup. Purification was achieved by flash chromatography (80 g silica-gel) in CH $_2$ Cl $_2$ /MeOH 97:3 to give a colourless foam. Yield 1.82 g (80 %). R $_f$ = 0.76 (CHCl $_3$ /MeOH 19:1).

Anal. calc. for $C_{18}H_{19}BrN_{2}O_{7}$ (455.3): C, 47.49; H, 4.21; N, 6.15. Found: C, 47.24; H, 4.38; N, 5.98.

 $\frac{1-(3,5-\text{Di-}0-\text{acetyl-}2-\text{chloro-}2-\text{deoxy-}\beta-\text{D-erythro-pento-furanosyl})-2,4-\text{quinazolinedione}}{(4.41\text{ g, }15\text{ mmole})\text{ and LiCl}} (4.24\text{ g, }100\text{ mmole})\text{ in CH}_3\text{CN}} (150\text{ ml})\text{ was heated to boiling and then a solution of acetyl chloride}} (7.85\text{ g, }100\text{ mmole})\text{ in CH}_3\text{CN}} (10\text{ ml})\text{ was added drop-wise. It was refluxed for 4 h, then evaporated and the residue taken up in CHCl}_3 (300\text{ ml})\text{ and saturated NaHCO}_3\text{ solution}} (200\text{ ml})\text{. The organic phase was dried, concentrated and purified by flash chromatography on silica-gel in CH}_2\text{Cl}_2/\text{EtOAc}} 4:1\text{. The main fraction gave on evaporation in vacuum an amorphous solid. Yield 4.49 g (75 %). R}_f = 0.79 (CHCl}_3/\text{MeOH 19:1})\text{.}}$

Anal. calc. for $C_{17}H_{17}C1\ N_2O_7\ (396.8)$: C, 51.46; H, 4.32; N, 7.06. Found: C, 51.33; H, 4.37; N, 7.02.

 $\frac{6-\text{Methyl-1-(3,5-di-0-acetyl-2-chloro-2-deoxy-$\beta-D-erythro-pentofuranosyl)-2,4-quinazolinedione}{2(36).-} \text{Analogous to the preceding procedure was reacted 2 (4.62 g, 15 mmole) with LiCl (4.24 g, 100 mmole) and acetyl chloride (7.85 g, 100 mmole) to give an amorphous solid. Yield 4.28 g (69 %). <math display="block">R_f = 0.49 \text{ (CHCl}_3/\text{EtOAc 4:1)}.$

Anal. calc. for $C_{18}H_{19}Cl$ N_2O_7 (410.8): C, 52.63; H, 4.66; N, 6.82. Found: C, 52.55; H, 4.63; N, 6.62.

A second fraction was also isolated and identified as $6-methyl-1-(2,3,5-tri-0-acetyl-\beta-D-ribofuranosyl)-2,4-quina-$

zolinedione. Yield 1.27 g (20 %). $R_f = 0.29$ (CHCl₃/EtOAc 4:1).

Anal. calc. for $C_{20}H_{22}N_{20}$ (434.4): C, 55.30; H, 5.10; N, 6.45. Found: C, 55.18; H, 5.17; N, 6.49.

 $\frac{1-(3,5-\text{Di}-0-\text{acetyl}-2-\text{deoxy}-\beta-D-\text{erythro-pentofuranosyl})-2,4-\text{quinazolinedione}}{2,4-\text{quinazolinedione}} \left(\frac{37}{2}\right).$ A solution of $\frac{35}{2}$ (3.97 g, 10 mmole) inTHF (100 ml) was purged by argon. Then tri-n-butylstannane (8 ml, 30 mmole) and AIBN (0.2 g) were added and the mixture heated to 65°C for 6 h under argon atmosphere. It was evaporated and the residue recrystallized from n-hexane to give colourless crystals. Yield 2.97 g (82 %), m.p. 120°C.

 $\frac{\text{Anal}}{\text{N, 7.73.}}$ calc. for $\text{C}_{17}\text{H}_{18}\text{N}_2\text{O}_7$ (362.3): C, 56.35; H, 5.01; N, 7.73. Found: C, 56.28; H, 5.11; N, 7.68.

<u>6-Methyl-1-(3,5-di-0-acetyl-2-deoxy-β-D-erythro-pento-furanosyl)-2,4-quinazolinedione</u> (<u>38</u>). - (a) A solution of <u>34</u> (0.71 g, 1.56 mmole) in toluene (50 ml) was gased with argon for 20 min and then tri-n-butylstannane (1.24 ml, 4.7 mmole) and AIBN (60 mg) were added. The mixture was heated to 75°C for 3 h under Ar atmosphere, then evaporated and the residue recrystallized from n-hexane to give colourless crystals. Yield 0.22 g (37 %).

- (b) Analogous to the receding prodedure $\underline{36}$ (1.51 g, 3.65 mmole) was heated in toluene (70 ml) with tri-n-butyl-stannane (2.92 ml, 11 mmole) and AIBN (0.1 g). Colourless crystals were obtained. Yield 1.03 g (75 %).
- (c) A solution of $\underline{24}$ (1.17 g, 4 mmole) in pyridine (10 ml) was treated at room temperature with acetic anhydride (10 ml) overnight. The reaction solution was poured onto ice and the resulting precipitate collected. Recrysrallization from EtOH/H $_2$ O gave colourless crystals. Yield 1.35 g (90 %), m.p. 170°C. $R_f=0.65$ (CHCl $_3$ /MeOH 19:1).

Anal. calc. for $C_{18}^{H}_{20}^{N}_{20}^{0}_{7}$ (376.4): C, 57.44; H, 5.36; N, 7.44. Found: C, 57.12; H, 5.48; N, 7.32.

 $\frac{1-(2-\text{Chloro-}2-\text{deoxy-}3,5-\text{di-}0-\text{pivaloyl-}\beta-\text{D-ribofurano-syl})-2,4-\text{quinazolinedione}}{\text{Syl})-2,4-\text{quinazolinedione}} \left(\frac{39}{2}\right).$ A mixture of $\frac{1}{2}$ (1.47 g, 5 mmole), LiCl (0.43 g, 10 mmole) and pivaloyl chloride (6.15 ml, 50 mmole) in CH₃CN (20 ml) was refluxed for 18 h. The suspension was then evaporated and the residue treated with CH₂Cl₂ (150 ml) and saturated NaHCO₃ solution (100 ml). The organic layer was dried over Na₂SO₄, filtered, evaporated and the residue purified by flash chromatography (silica-gel column 4.5x14 cm) with CH₂Cl₂/Et0Ac 9:1 to give a colourless amorphous solid. Yield 1.17 g (49 %). R_f = 0.39 (CH₂Cl₂/Et0Ac 9:1).

 $\frac{\text{Anal}}{\text{Anal}}$. calc. for $\text{C}_{23}\text{H}_{29}\text{Cl N}_2\text{O}_7$ (480.9): C, 57.44; H, 6.08; N, 5.82. Found: C, 57.18; H, 6.12; N, 5.44.

 $\frac{6-\text{Chloro}-1-(2-\text{chloro}-2-\text{deoxy}-3,5-\text{di}-0-\text{pivaloyl}-\beta-D-\text{ri}-\text{bofuranosyl})-2,4-\text{quinazolinedione}}{\text{bofuranosyl}-2,4-\text{quinazolinedione}} \left(\frac{40}{1}\right). - \text{Analogous to the preceding procedure was reacted } \frac{6}{1} \left(1.64 \text{ g, 5 mmole}\right) \text{ to give a colourless foam. Yield 2.45 g (95 %). R}_{f} = 0.77 \text{ (CHCl}_{3}/-\text{EtOAc 4:1}).}$

 $\frac{\text{Anal}}{\text{Anal}}$. calc. for $\text{C}_{23}\text{H}_{28}\text{Cl}$ N_2O_7 (515.4): C, 53.60; H, 5.48; N, 5.44. Found: C, 54.13; H, 5.85; N, 5.18.

 $\frac{1-(2-\text{Deoxy}-3,5-\text{di}-0-\text{pivaloyl}-\beta-D-\text{erythro-pentofurano-syl})-2,4-\text{quinazolinedione}}{\text{syl})-2,4-\text{quinazolinedione}} \ (41). - \text{A solution of } \frac{39}{39} \ (0.96 \text{ g}, 2 \text{ mmole}) \ \text{in toluene}} \ (20 \text{ ml}) \ \text{was gased with Ar for 20 min.}$ Then tri-n-butylstannane (1.6 ml, 6 mmole) and AIBN (50 mg) were added and the mixture heated to 75°C for 6 h under Ar atmosphere. After evaporation the residue was heated in n-hexane till crystallization took place. Colourless crystals. Yield 0.67 g (75%), m.p. 156°C. R_f = 0.61 (CHCl₃/EtOAc 4:1).

Anal. calc. for $C_{23}H_{30}N_2O_7$ (446.5): C, 61.87; H, 6.77; N, 6.27. Found: C, 61.66; H, 6.70; N, 6.37.

6-Chloro-1-(2-deoxy-3,5-di-0-pivaloyl- β -D-erythro-pento-furanosyl)-2,4-quinazolinedione (42). - Analogous to the preceding procedure with 40 (1.03 g, 2 mmole). The crude pro-

duct could not be crystallized. Purification was achieved by flash chromatography (silica-gel column 2.5x15 cm) in ${\rm CH_2Cl_2/-EtOAc}$ 4:1 to give a colourless foam. Yield 0.77 g (80 %). ${\rm R_f}=0.58$ (CHCl $_3/{\rm EtOAc}$ 4:1).

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Anal.calc. for $C_{23}H_{29}C1$ N_2O_7 (480.9): C, 57.44; H, 6.08; N, 5.82. Found: C, 57.52; H, 6.30; N, 5.99.

1-(2-0-Acetyl-3-deoxy-5-0-p-toluoyl-β-D-erythro-pentofuranosyl)-2,4-quinazolinedione (49-53). General procedure. A mixture of the 2,4-quinazolinedione (43-47) (2 mmole), a few crystals of ammonium sulfate and acetamide was refluxed in hexamethyldisilazane (HMDS) (20 ml) for 20 h under exclusion of moisture. The excess of HMDS was removed in vacuum and the residue dissolved in a solution of 1,2-di-0-acety1-3deoxy-5-0-p-toluoyl-D-ribose $(48)^{38}$ in 1,2-dichloroethane (60 ml). A solution of trimethylsilyl trifluoromethanesulfonate (0.44 ml, 2.2 mmole) in 1,2-dichloroethane (10 ml) was added dropwise with stirring at room temperature. The reaction was stopped by addition of a saturated solution of Na- HCO_3 and then extracted twice with $CHCO_3$ (80 ml). The combined organic layers were dried over Na₂SO₄, filtered, evaporated and the residue purified by flash chromatography. The main fraction gave an amorphous solid on evaporation.

 $\frac{1-(2-0-Acetyl-3-deoxy-5-0-p-toluoyl-\beta-D-erythro-pento-furanosyl)-2,4-quinazolinedione}{(49)}. - Flash chromatography (25 g silica-gel) was performed in CHCl3. Yield 0.52 g (59 %). <math>R_f = 0.30$ (CHCl3/EtOAc 4:1).

Anal. calc. for $C_{23}H_{22}N_2O_7$ (438.4): C, 63.01; H, 5.06; N, 6.39. Found: C, 62.65; H, 5.16; N, 6.20.

 $\frac{6-\text{Methyl-1-}(2-0-\text{acetyl-3-deoxy-5-0-p-toluoyl-}\beta-D-\text{erythro-pentofuranosyl})-2,4-\text{quinazolinedione}}{\text{done by flash chromatography in CHCl}_3/\text{MeOH 99:1. Yield 0.56 g}}$

- Anal. calc. for $C_{24}H_{24}N_2O_7$ (452.4): C, 63.71; H, 5.35; N, 6.19. Found: C, 63.67; H, 5.38; N, 6.19.
- $\frac{7-\text{Methyl-1-}(2-0-\text{acetyl-3-deoxy-5-0-p-toluoyl-}\beta-D-\text{erythro-pentofuranosyl})-2,4-\text{quinazolinedione}}{(51)}.-\text{Analogous to the preceding procedure gave an amorphous foam. Yield 0.61 g}}{(67 \%). R_f=0.33 (CHCl_3/Et0Ac 4:1).}$
- Anal. calc. for $C_{24}H_{24}N_2O_7$ (452.4): C, 63.71; H, 5.35; N, 6.19. Found: C, 63.61; H, 5.53; N, 6.25.
- $\frac{6.7-\text{Dimethyl}-1-(2-0-\text{acetyl}-3-\text{deoxy}-5-0-\text{p-toluoyl}-\beta-D-\text{erythro-pentofuranosyl})-2.4-\text{quinazolinedione}}{\text{cation by flash chromatography in CHCl}_3/\text{MeOH 99:1. Yield}}$ $0.68 \text{ g (73 \%). R}_f = 0.45 \text{ (CHCl}_3/\text{EtOAc 4:1).}$
- <u>Anal.</u> calc. for $C_{25}H_{26}N_2O_7$. 0.5 H_2O (475.5): C, 63.15; H, 5.72; N, 5.89. Found: C, 63.52; H, 5.71; N, 5.74.
- $\frac{6.7-\text{Dimethoxy-1-}(2-0-\text{acetyl-3-deoxy-5-0-p-toluoyl-}\beta-D-\text{erythro-pentofuranosyl})-2.4-\text{quinazoline-dione}}{\text{chromatography in CHCl}_3/\text{MeOH }99:1 \text{ gave a solid foam. Yield}}\\ 0.67 \text{ g } (68 \text{ \%}). \text{ R}_f = 0.34 \text{ (CHCl}_3/\text{EtOAc }4:1).}$
- $\frac{\text{Anal}}{\text{Anal}}$. calc. for $\text{C}_{25}\text{H}_{26}\text{N}_2\text{O}_9$. H_2O (516.5): C, 58.14; H, 5.46; N, 5.42. Found: C, 58.33; H, 5.17; N, 5.28.
- $\frac{1-(3-\text{Deoxy}-\beta-\text{D-erythro-pentofuranosyl})-2,4-\text{quinazoline-diones}}{\text{diones}} (\frac{54-58}{2}). \quad \underline{\text{General procedure}}. \quad \quad \text{To a solution of sodium} (26 \text{ mg}, 1.1 \text{ mmole}) \text{ in MeOH} (20 \text{ ml}) \text{ was added the appropriate protected quinazoline } 3'-\text{deoxynucleoside} \frac{49-53}{2} (1 \text{ mmole}). \quad \text{The solution was stirred at room temperature overnight, then evaporated and the residue partitioned between } H_20 (2x10 \text{ ml}) \text{ and ether } (50 \text{ ml}). \quad \text{The combined aqueous layers were neutralized by acetic acid and on chilling overnight colourless crystals were obtained. The solid was collected, washed and dried at <math>100^{\circ}\text{C}$.

- $\frac{1-(3-\text{Deoxy-}\beta-\text{D-erythro-pentofuranosyl})-2,4-\text{quinazoline-dione}}{\text{dione}} (54). \text{Yield 0.16 g (56 \%), m.p. } 194-195^{\circ}\text{C}.$
- Anal. calc. for $C_{13}H_{14}N_2O_5$ (278.3): C, 56.11; H, 5.07; N, 10.07. Found: C, 55.91; H, 5.10; N, 9.90.
- 6-Methyl-1-(3-deoxy- β -D-erythro-pentofuranosyl)-2,4-quinazolinedione (55). Yield 0.24 g (80 %), m.p. 232°C.
- Anal. calc. for $C_{14}H_{16}N_2P_5$ (292.3): C, 57.33; H, 5.52; N, 9.58. Found: C, 57.54; H, 5.58; N, 9.52.
- $7-Methyl-1-(3-deoxy-\beta-D-erythro-pentofuranosyl)-2,4-quinazolinedione (56). Yield 0.22 g (74 %), m.p. 230°C.$
- Anal. calc. for $C_{14}H_{16}N_{2}O_{5}$ (292.3): C, 57.53; H, 5.52; N, 9.58. Found: C, 57.33; H, 5.58; N, 9.77.
- $\frac{6.7-\text{Dimethyl-1-}(3-\text{deoxy-}\beta-D-\text{erythro-pentofuranosyl})}{2.4-\text{quinazolinedione}} = \frac{57}{221^{\circ}C}.$ Yield 0.22 g (71 %), m.p. 219-221°C.
- Anal. calc. for $C_{15}H_{18}N_2O_5$ (306.3): C, 58.84; H, 5.56; N, 9.15. Found: C, 58.81; H, 5.95; N, 8.93.
- $\frac{6,7-\text{Dimethoxy-1-}(3-\text{deoxy-}\beta-\text{D-erythro-pentofuranosy1})-2,4-\text{quinazolinedione}}{2,4-\text{quinazolinedione}} (\underline{58}). \text{Yield 0.22 g (65 \%), m.p. 207-209°C}.$
- Anal. calc. for $C_{15}H_{18}N_2O_7$ (338.3): C, 53.25; H, 5.36; N, 8.28. Found: C, 52.84; H, 5.45; N, 8.29.
- $\frac{1-(2,3-\text{Dideoxy-}5-0-\text{p-toluoyl-}\beta-\text{D-glycero-pentofuranosyl})-2,4-\text{quinazolinediones}}{2,4-\text{quinazolinediones}} (59-62). \underline{\text{General procedure}}. A suspension of the appropriate 1-(5-0-\text{p-toluoyl-}\beta-\text{D-ribofuranosyl})-2,4-\text{quinazolinedione}} (29-32) (1.5 mmole) and pyridine (0.21 ml, 2.6 mmole) in <math>\text{CH}_2\text{Cl}_2$ (20 ml) was cooled to 0°C and then phenoxythiocarbonyl chloride (0.35 ml, 2.6 mmole) added dropwise with stirring. The reaction mixture was kept 3 h at

0°C and then at room temperature overnight. After evaporation the residue was partitioned between $\rm H_2O$ (30 ml) and EtOAc (2x30 ml). The combined organic layers were dried over $\rm Na_2-SO_4$, filtered, evaporated and the remaining syrup dissolved in toluene/THF 1:1 (50 ml). The solution was treated with argon for 20 min and after addition of tri-n-butylstannane (1.2 ml, 4.5 mmole) and AIBN (50 mg) heated to 75°C for 3 h with stirring under Ar atmosphere. After cooling the solution was evaporated and the residue refluxed in n-hexane (30 ml) till crystallization took place. The crystals were collected and dried in vacuum at 40°C.

 $\frac{1-(2,3-\text{Dideoxy-}5-0-\text{p-toluoy1-}\beta-\text{D-glycero-pentofurano-syl})-2,4-\text{quinazolinedione}}{\text{m.p.} 94-96°C.} = 0.27 (CHCl_3/EtOAc 4:1).$

 $\frac{\text{Anal.calc. for C}_{21}\text{H}_{20}\text{N}_{2}\text{O}_{5}\text{ . 0.5 H}_{2}\text{O (380.4): C, 64.77;}}{\text{H, 5.43; N, 7.19. Found: C, 64.56; H, 5.50; N, 7.45.}}$

 $\frac{6-\text{Methyl-1-}(2,3-\text{dideoxy-5-0-p-toluoyl-}\beta-D-\text{glycero-pento-furanosyl})-2,4-\text{quinazolinedione}}{\text{m.p. }162-165^{\circ}\text{C. }R_{f}=0.67\text{ (CHCl}_{3}/\text{MeOH }19:1).}$

Anal. calc. for $C_{22}H_{22}N_2O_5$. 0.5 H_2O (403.4): C, 65.50; H, 5.75; N, 6.94. Found: C, 65.00; H, 5.73; N, 6.57.

 $\frac{7-\text{Methyl-1-}(2,3-\text{dideoxy-5-0-p-toluoyl-}\beta-D-\text{glycero-pento-furanosyl})-2,4-\text{quinazolinedione}}{\text{m.p. 149°C. }R_f=0.32 \text{ (CHCl}_3/\text{EtOAc 4:1).}}$

Anal. calc. for $C_{22}H_{22}N_2O_5$. 0.5 H_2O (403.4): C, 65.50; H, 5.75; N, 6.94. Found: C, 65.48; H, 5.70; N, 6.85.

 $\frac{6.7-\text{Dimethyl-1-}(2.3-\text{dideoxy-5-0-p-toluoyl-}\beta-\text{D-glycero-pentofuranosyl})-2.4-\text{quinazolinedione}}{\text{(89 \%), m.p. }175^{\circ}\text{C. R}_{f}=0.30\text{ (CHCl}_{3}/\text{EtOAc 4:1).}}$

Anal. calc. for $C_{23}H_{24}N_2O_5$. H_2O (426.5): C, 64.78; H, 6.15; N, 6.57. Found: C, 65.14; H, 6.20; N, 6.64.

- $\frac{1-(2,3-\text{Dideoxy-}\beta-\text{D-glycero-pentofuranosyl})-2,4-\text{quinazo-linediones}}{\text{linediones}} (63-66). \underline{\text{General procedure}}. \text{The protected quinazoline 2',3'-dideoxyribonucleoside} (\underline{59-62}) (1 \,\text{mmole})$ was stirred at room temperature overnight in a solution of sodium (28 mg, 1.2 mmole) in MeOH (50 ml). The reaction mixture was evaporated and the residue partitioned between ether (20 ml) and H₂O (2x5 ml). The combined aqueous layers were neutralized by AcOH and after cooling the precipitated crystals collected, washed and dried.
- $1-(2,3-Dideoxy-\beta-D-glycero-pentofuranosyl)-2,4-quinazo-linedione (63). Yield 0.2 g (76 %), m.p. 175°C.$
- Anal. calc. for $C_{13}H_{14}N_2O_4$ (262.3): C, 59.54; H, 5.38; N, 10.68. Found: C, 59.88; H, 5.43; N, 10.63.
- $\frac{6-\text{Methyl-1-}(2,3-\text{dideoxy-}\beta-D-\text{glycero-pentofuranosyl})}{2,4-\text{quinazolinedione}} \frac{2,4-\text{quinazolinedione}}{2,4-\text{quinazolinedione}} \frac{6-\text{Methyl-1-}(2,3-\text{dideoxy-}\beta-D-\text{glycero-pentofuranosyl})}{2,4-\text{quinazolinedione}} \frac{6-\text{Methyl-1-}(2,3-\text{dideoxy-}\beta-D-\text{glycero-pentofuranosyl})}{2,4-\text{quinazoli$
- Anal. calc. for $C_{14}H_{16}N_2O_4$. 0.5 H_2O (285.3): C, 58.94; H, 6.00; N, 9.82. Found: C, 59.20; H, 5.77; N, 9.93.
- $\frac{7-\text{Methyl-1-(2,3-dideoxy-}\beta-D-glycero-pentofuranosyl)}{2,4-quinazolinedione~(65).-Yield~0.232~g~(84~\%),~m.p.~198°C.}$
- Anal. calc. for $C_{14}H_{16}N_{2}O_{4}$ (276.3): C, 60.85; H, 5.84; N, 10.14. Found: C, 60.40; H, 5.92; N, 10.18.
- $\frac{6.7-\text{Dimethyl-1-(2,3-dideoxy-}\beta-D-\text{glycero-pentofuranosyl)}}{2.4-\text{quinazoliendione}} \left(\frac{66}{6}\right). \text{Yield 0.25 g (85 \%), m.p. 220°C} \left(\frac{66}{6}\right).$
- $\underline{\text{Anal}}$. calc. for $C_{15}H_{18}N_2O_4$. 0.25 H_2O (294.8): C, 61.11; H, 6.33; N, 9.50. Found: C, 61.04; H, 6.25; N, 9.29.

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